

Name  
TA

Chemistry 11  
2021-2022



## Learning Guide # 7: Quantum Theory and Electronic Structure

**BIG IDEA:** Quantum Mechanics, Atomic Orbitals, and Electron Configuration

### Fundamental Knowledge (I know)

- The features of a wave AND how to calculations dealing with Wave Length
- The different electron orbitals their associated energy levels (aka, which level to assign first, second, third, etc.)
- How to correctly assign an element's electrons to the correct orbital

### Curricular Competencies (I can)

	Proficiency Scale Teacher and Student self assessment (Circle one)	Evidence (How do you know?)
<b>I can:</b>  Seek and analyze patterns, trends, and connections in data, including describing relationships between variables, performing calculations, and identifying inconsistencies.	<b>Emerging (EMG)</b> Initial Understanding  <b>Developing (DEV)</b> Partial/Near Complete Understanding  <b>Proficient (PRF)</b> Complete Understanding  <b>Extending (EXT)</b> Sophisticated Understanding	
Construct, analyze, and interpret graphs, models, and/or diagrams	<b>Emerging (EMG)</b> Initial Understanding  <b>Developing (DEV)</b> Partial/Near Complete Understanding  <b>Proficient (PRF)</b> Complete Understanding  <b>Extending (EXT)</b> Sophisticated Understanding	

Student Signature:

Teacher Signature:

Date:

**Instructions** To help guide your learning, make your way through the activities in Option 1, Option 2, or Option 3. You may “mix and match” between the different Option columns.

TOPIC	OPTION 1	OPTION 2	OPTION 3
History of Quantum Theory	<p><b>Create</b> a glossary of the “Key Words” in chapter 7 (Just 7.1, 7.5, 7.7.7.8, and 7.9) (Pgs. 268 – 272, 283 – 286, 288 - 302)</p> <p><b>Read</b> Pages 268 – 273 and <b>complete Review Questions:</b> 7.1, 7.2, 7.3, and 7.5 on Pg. 303.</p> <p><b>Complete “Example: Practice Exercises”</b> 7.1 and 7.2 on Pgs. 269 and 270.</p>	<p><b>Create</b> five (5) questions similar to “Example” 7.1 and 7.2.</p> <p><b>Complete Review Questions:</b> 7.1, 7.2, 7.3, and 7.5 on Pg. 303.</p> <p><b>Complete “Example: Practice Exercises”</b> 7.1 and 7.2 on Pgs. 269 and 270.</p>	<p><b>Choose your own adventure!</b></p> <p>Pick up a planning sheet from the Science Kiosk.</p> <p>Create a plan! Make sure you read through the first page of this LG, as you will need to design ways to learn/practice and show your understanding of the topic(s) and skill(s) (competencies.)</p> <p>You will need to have a teacher approve your plan before beginning the LG.</p>
Quantum Mechanics	<p><b>Read</b> Pages 274 – 283 and <b>complete Review Questions:</b> 7.43 and 7.44, on Pg. 305.</p>	<p><b>Write</b> a paragraph summarizing the information about quantum mechanics.</p> <p><b>Complete Review Questions:</b> 7.43 and 7.44, on Pg. 305.</p>	
Atomic Orbitals	<p><b>Read</b> Pages 288 – 292 and <b>complete Review Questions:</b> 7.48, on Pg. 305.</p>	<p><b>Create</b> a digital presentation explaining how to draw the electron configuration of ground state elements. Use visuals and examples in your presentation to show your learning.</p> <p><b>Complete Review Questions:</b> 7.48, on Pg. 305.</p>	
Electron Configuration	<p><b>Read</b> Pages 292 – 298 and <b>complete Review Questions:</b> 7.71, 7.72, and 7.73 on Pg. 306.</p>	<p><b>Complete Review Questions:</b> 7.71, 7.72, and 7.73 on Pg. 306.</p>	
The Build Up Principle	<p><b>Read</b> Pages 298 – 302 and <b>complete Review Questions:</b> 7.81 on Pg. 306.</p> <p><b>Complete “Example: Practice Exercises”</b> 7.11 on Pgs. 301 and 302.</p>	<p><b>Complete Review Questions:</b> 7.81 on Pg. 306.</p> <p><b>Complete “Example: Practice Exercises”</b> 7.11 on Pgs. 301 and 302.</p>	
<b>Chapter Review</b>	<b>Complete “problems” 7.8, 7.65, 7.76, 7.90, and 7.114</b>		
Lab	NO LAB		
Self Assessment	Reflect on the Fundamental Knowledge and Curricular Competencies. Use the rubric and make goals to improve for your next learning guide.		
Interview or Quiz	See you teacher for an interview or to have a quiz slip signed for the test center. Bring your work and staple it to your quiz when complete.		

Resources can be found at [www.THSSscience.com](http://www.THSSscience.com) or the Science Kiosk

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