

## **Learning Guide # 2: Atoms, Molecules and Ions**

BIG IDEA: Parts of The Atom, the Periodic Table, and Bonding

Fundame	ental Knowledge (I know)				
	How to properly name an randomly chosen Ionic Compound				
	he History that lead to the current view of the atomic structure				
	ne difference between an Ion and an Atom				
	ow to correctly write symbols for isotopes that show number of protons and neutrons				
	low to convert a chemical/empirical formula into a molecular formula				
	low the Periodic table is organized AND can explain the difference groups, rows, and trends				
	How to properly convert the n	ame of a molecular compound to its formula and vice versa			
Curricula	r Competencies (I can)				
Curricula	Proficiency Scale				
	Teacher and Student self	Evidence			
	assessment				
	(Circle one)	(How do you know?)			
I can:	Emerging (EMG)				
	Initial Understanding				
Seek and					
analyze	Developing (DEV)				
patterns,	Partial/Near Complete				
trends, and connections in	Understanding				
data, including					
describing	Proficient (PRF)				
relationships	Complete Understanding				
between					
variables,					
performing calculations,	Extending (EXT)				
and identifying	Sophisticated Understanding				
inconsistencies.					
	Emerging (EMG)				
Consider the	Initial Understanding				
changes in	Developing (DEV)				
knowledge over	Partial/Near Complete				
time as tools	Understanding Proficient (PRF)				
and technologies	Complete Understanding				
have developed.					
3.2.2.2.2.0peu.	Sophisticated Understanding				

**Instructions** To help guide your learning, make your way through the activities in Option 1, Option 2, or Option 3. You may "mix and match" between the different Option columns.

TOPIC	OPTION 1	OPTION 2	OPTION 3	
Atomic Theory and Structure of the Atom	Create a glossary of the bolded terms in chapter 2 (Pgs. 42 – 69)  Read Pages 42 – 48 and complete Review Questions: 2.1 - 2.6 on Pg. 70.	Create a digital presentation outline the parts of the Atom and summarizing the Scientists and discoveries that lead to our current atomic theory.	Choose your own adventure!  Pick up a planning sheet from the Science Kiosk.	
		Read Pages 42 – 48 and complete Review Questions: 1.1 - 1.6 on Pg. 70.	Create a plan! Make sure you read through the first	
The Periodic Table: Atomic Number, Atomic Mass, and Isotopes	Read Pages 49 – 51 and complete Review Questions: 2.9 - 2.12 on Pg. 71.  Complete "Example: Practice Exercises" 2.1 on pg 50.	Find and SOURCE a video and or link that explains isotopes and complete Review Questions: 2.9 - 2.12 on Pg. 71.  Complete "Example: Practice Exercises" 2.1 on pg 50.	page of this LG, as you will need to design ways to learn/practice and show your understanding of the topic(s) and skill(s)	
Molecules, lons, and Chemical Formulas	Read Pages 53 – 58 and complete Review Questions: 2.19 - 2.22, 2.27 – 2.30 on Pg. 71.  Complete "Example: Practice Exercises" 2.2, 2.3, and 2.4 on pgs 56 and 58.	Create step by step notes that explain how to convert a chemical/empirical formula into a molecular formula.  Complete Review Questions: 2.19 - 2.22, 2.27 - 2.30 on Pg. 71.  Complete "Example: Practice Exercises" 2.2, 2.3, and 2.4 on pgs 56 and 58.	(competencies.)  You will need to have a teacher approve your plan before beginning the LG.	
Naming Chemical Compounds	Read Pages 59 – 69 and complete Review Questions: 2.37 - 2.4, 2.51 – 2.56 on Pg. 72.  Complete "Example: Practice Exercises" 2.5-2.9 on Pg. 61 – 66.	Create a digital presentation summarizing the information on pages 59 – 69.  Complete Review Questions: 2.37 - 2.4, 2.51 – 2.56 on Pg. 72.  Complete "Example: Practice Exercises" 2.5-2.9 on Pg. 61 – 66.		
Chapter Review	Complete "problems" 2.16, 2.24, 2.26, 2.36, 2.47, 2.48, 2.50, 2.58, 2.60, 2.63, 2.68, 2.89 on pgs.71–74.			
Lab	Lab 3.1A: Recognizing elements, compounds and mixtures.			
Self Assessment	Reflect on the Fundamental Knowledge and Curricular Competencies. Use the rubric and make goals to improve for your next learning guide.			
Interview or Quiz	See you teacher for an interview or to have a quiz slip signed for the test center. Bring your work and staple it to your quiz when complete.			